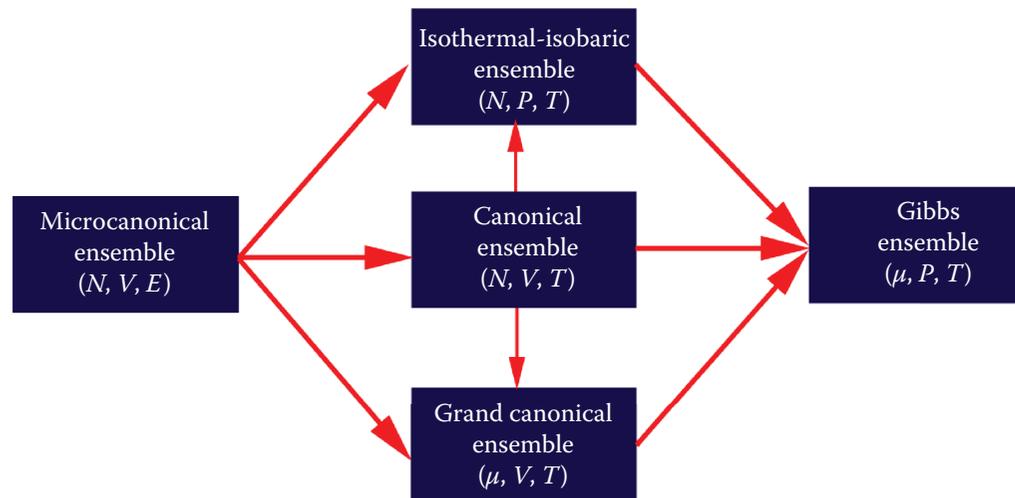


# Ensemble

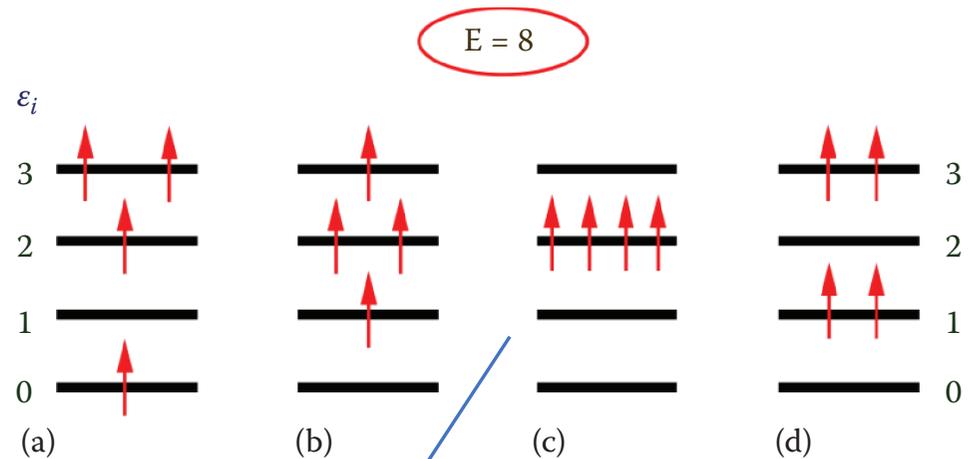
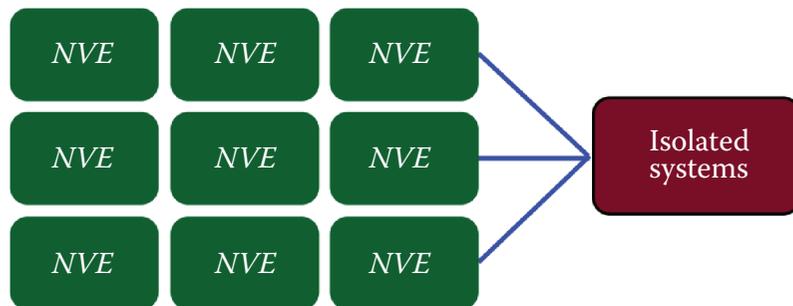
The concept of an ensemble is a **brilliant mental construct**

System must have a large number of microscopic states (positions and momenta), and natural motion of system at non-zero temperature takes the system through a finite fraction of these states in a time comparable to time of measurement of the macroscopic properties.

## Different types of Ensembles

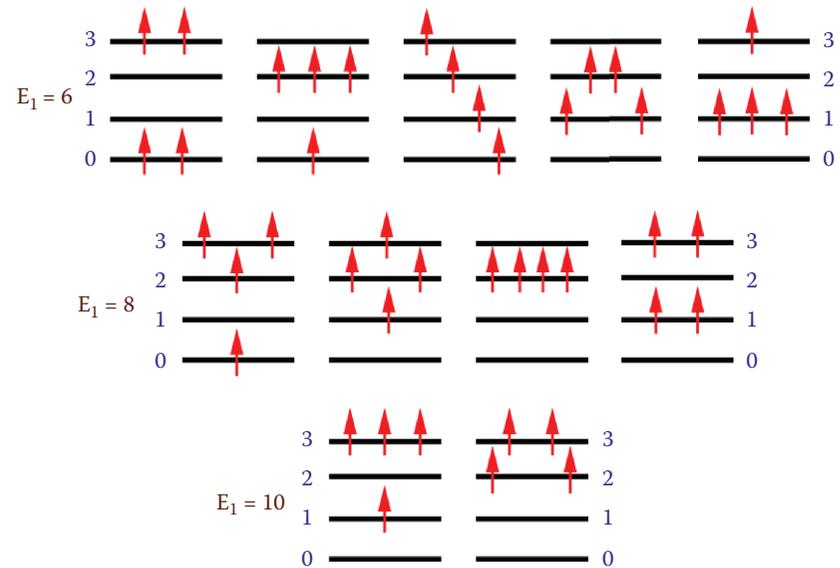
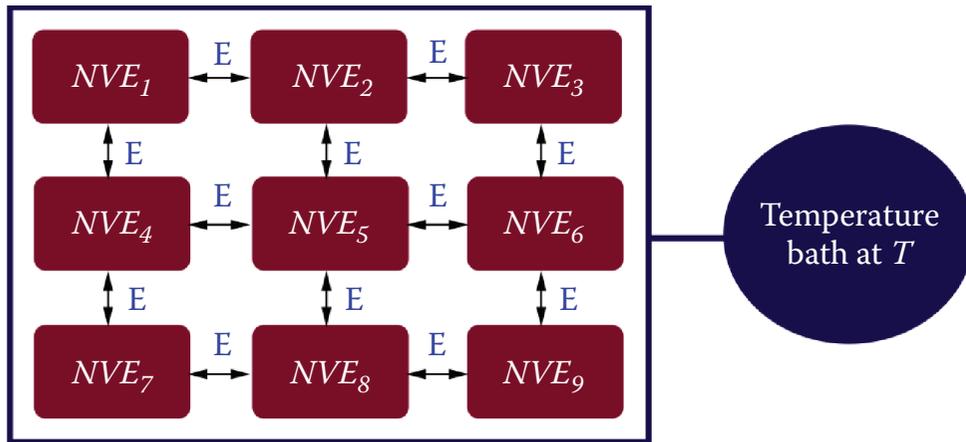


A **microcanonical ensemble** consists of mental replicas of the original (NVE) system (**Isolated System**)



Four arrangements giving rise to same total energy (these are microstates)

## Canonical ensemble



Canonical ensemble, where energy of each system can fluctuate. The systems are kept at temperature  $T$  by putting the super-system in a bath and establishing thermal contacts between the individual systems

$N_s \rightarrow$  Total no. of Systems

$N \rightarrow$  No. of particles in each System

Super-System  
(Isolated)

$N_t = N_s N$   
 $\rightarrow$  total no. of particles

$V_t = N_s V$   
 $\rightarrow$  total volume

$E_t =$  Total energy

$(N_t, V_t, E_t)$

then we  
can construct  
a microcanonical  
ensemble starting  
here

$$\sum_j n_j = N_s$$

$$\sum_j n_j E_j = E_t$$

$\rightarrow$  constraints

If the particles are distinguishable

$$\Omega(\{n_j\}) = \frac{N_s!}{n_1! n_2! n_3! \dots} \quad (\text{elaborate this later}) \dots \textcircled{1}$$

↘ No. of ways .....

x Biman Bagchi }  
x Mc. Quarrie }

Probability of observing a given state  $n_j$  with energy  $E_j$

$$p_j = \frac{\bar{n}_j}{N_s} = \frac{1}{N_s} \left( \frac{\sum_j n_j \Omega(\{n_j\})}{\sum_j \Omega(\{n_j\})} \right) \rightarrow \bar{n}_j \dots \textcircled{2}$$

[In general, there are many distributions that are consistent with  $\sum_j n_j = N_s$ .]

Most probable value

$n_j^*$  → is the value of  $n_j$  that maximises  $\Omega(\{n_j\})$

$$\approx \frac{1}{N_s} \frac{n_j^* \Omega(\{n_j^*\})}{\Omega(\{n_j^*\})} = \frac{n_j^*}{N_s}$$

In any particular distribution,  $\frac{n_j}{N_s}$  is

the fraction of systems of the canonical ensemble

in the  $j$ -th energy state. The overall probability

$p_j$  that a system is in the  $j$ -th energy state is

obtained by averaging  $\frac{n_j}{N_s}$  over all the allowed distributions, giving equal weight to the each one  
principle of equal a priori probabilities

$$\frac{\partial}{\partial n_j} \left( \ln \Omega(\{n_j\}) - \alpha \sum_j n_j - \beta \sum_j n_j E_j \right) = 0$$

$\swarrow$  LUM  $\searrow$

$$\left( 0 - \frac{1}{n_j} - \ln n_j^* - \alpha - \beta E_j \right) = 0$$

$$-1 - \ln n_j^* - \alpha - \beta E_j = 0$$

$$\text{or, } \ln n_j^* = -(1 + \alpha) - \beta E_j \quad \alpha' = 1 + \alpha$$

$$= -\alpha' - \beta E_j$$

$$n_j^* = e^{-\alpha'} e^{-\beta E_j}$$

$$\sum n_j^* = N_s = e^{-\alpha'} \sum_j e^{-\beta E_j}$$

$$\text{or, } e^{+\alpha'} = \frac{\sum_j e^{-\beta E_j}}{N_s}$$

$\Rightarrow$

$$N_s = \sum n_j$$

$$\sum n_j E_j = E_t$$

$$\ln N! \approx N \ln N - N$$

(Stirling Approximation)

$$\ln \Omega(\{n_j\}) = \ln N_s! - \ln \prod_j n_j!$$

$$= N_s \ln N_s - N_s - \sum_j \ln n_j!$$

[  $n_1! n_2! \dots$  ]

$$= N_s \ln N_s - \frac{N_s}{s} - \sum_j n_j \ln n_j + \sum_j n_j$$

$\rightarrow N_s$

$$\ln \Omega = N_s \ln N_s - \sum_j n_j \ln n_j$$

$$P_j = \frac{n_j^*}{N_s} = \frac{N_s}{\sum_j e^{-\beta E_j}} \cdot \frac{e^{-\beta E_j}}{N_s}$$

$$P_j = \frac{e^{-\beta E_j}}{\sum_j e^{-\beta E_j}}$$

$$Q_N(v, T) = \sum_j e^{-\beta E_j}$$

Canonical partition function

$\rightarrow$  Boltzmann distribution

What about  $\beta$ ?  $\rightarrow$  LATER and show  $\beta = \frac{1}{k_B T}$

$$\bar{E} = \sum_j P_j E_j = \frac{\sum_j E_j e^{-\beta E_j}}{\sum_j e^{-\beta E_j}}$$

$$d\bar{E} = \sum_j (E_j dP_j + P_j dE_j)$$

$$P_j = \frac{e^{-\beta E_j}}{Q}, \quad \ln P_j = -\beta E_j - \ln Q$$

or,  $E_j = -\frac{1}{\beta} (\ln P_j + \ln Q)$

$$d\bar{E} = \sum_j \left( -\frac{1}{\beta} \right) (\ln P_j + \ln Q) dP_j + \sum_j P_j \left( \frac{\partial E_j}{\partial V} \right) dV$$

$$d\bar{E} = -\frac{1}{\beta} \sum_j \ln P_j dP_j - \frac{1}{\beta} \ln Q \sum_j dP_j + \sum_j P_j \left( \frac{\partial E_j}{\partial V} \right) dV$$

$$S = -k_B \sum_j P_j \ln P_j$$

$$dS = -k_B d \sum_j P_j \ln P_j$$

$$d\bar{E} = -\frac{1}{\beta} \sum_j \ln P_j dP_j + \sum_j P_j \left( \frac{\partial E_j}{\partial V} \right) dV$$

$$\beta = - \sum_j P_j \left( \frac{\partial E_j}{\partial V} \right) = -k_B \left( \sum_j \frac{P_j}{P_j} dP_j + \sum_j \ln P_j dP_j \right)$$

$$d\bar{E} = \left( \frac{1}{\beta} \right) \left( -\frac{dS}{k_B} \right) + \sum_j P_j \left( \frac{\partial E_j}{\partial V} \right) dV$$

Thermodynamics

$$dS = -k_B \sum_j \ln P_j dP_j$$

$$d\bar{E} = \frac{1}{\beta k_B} dS + \sum_j P_j \left( \frac{\partial E_j}{\partial V} \right) dV \quad \text{--- (A)}$$

$$dE = T dS - p dV \quad \text{--- (B)}$$

$$\Rightarrow T = \frac{1}{\beta k_B} \Rightarrow \beta = \frac{1}{k_B T}$$

$$Q = \sum_j e^{-\beta E_j}$$

$$\ln Q = \ln \left( \sum_j e^{-\beta E_j} \right)$$

$$\frac{\partial \ln Q}{\partial \beta} = \frac{1}{\sum_j e^{-\beta E_j}} \sum_j (-E_j) e^{-\beta E_j} = - \frac{\sum_j E_j e^{-\beta E_j}}{\sum_j e^{-\beta E_j}} = -\bar{E}$$

$$\bar{E} = \sum_j P_j E_j \quad P_j = \frac{e^{-\beta E_j}}{Q}$$

$$\beta = \frac{1}{k_B T}$$

$$d\beta = -\frac{1}{k_B T^2} dT$$

$$\text{or } \frac{\partial}{\partial \beta} = -k_B T^2 \frac{\partial}{\partial T}$$

$$\bar{E} = -\frac{\partial \ln Q}{\partial \beta}$$

$$\bar{E} = k_B T^2 \left( \frac{\partial \ln Q}{\partial T} \right)_{N, V}$$

Internal energy

$$\frac{\partial \ln Q}{\partial V} = \frac{1}{\sum_j e^{-\beta E_j}} (-\beta) \sum_j e^{-\beta E_j} \left( \frac{\partial E_j}{\partial V} \right)$$

$$= \beta \left( - \frac{\sum_j e^{-\beta E_j} (\partial E_j / \partial V)}{Q} \right) = \beta p \leftarrow \text{Pressure}$$

$$p = k_B T \left( \frac{\partial \ln Q}{\partial V} \right)_{N, T}$$

$$S = -k_B \sum_j P_j \ln P_j$$

$$= -k_B \sum_j \frac{e^{-\beta E_j}}{Q} \ln \left( \frac{e^{-\beta E_j}}{Q} \right) = -k_B \sum_j \frac{e^{-\beta E_j}}{Q} (-\beta E_j - \ln Q)$$

$$= k_B \beta \left( \sum_j \frac{e^{-\beta E_j} E_j}{Q} \right) + k_B \ln Q$$

$$S = \frac{1}{T} \bar{E} + k_B \ln Q = k_B T \left( \frac{\partial \ln Q}{\partial T} \right)_{N,V} + k_B \ln Q$$

~~$$A = k_B T^2 \left( \frac{\partial \ln Q}{\partial T} \right)_{N,V} - k_B T^2 \left( \frac{\partial \ln Q}{\partial T} \right)_{N,V} - k_B T \ln Q$$~~

$A = E - TS$   
 ↓  
 Helmholtz  
 free energy

$$A = -k_B T \ln Q$$

→ Thermodynamic potential  
 in canonical ensemble

$$S = k_B \ln \Omega$$

↳  
 T.P. in  
 microcanonical  
 ensemble

# Grand-canonical ensemble

